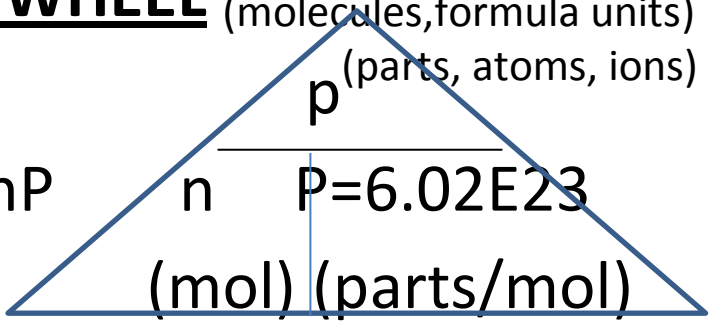
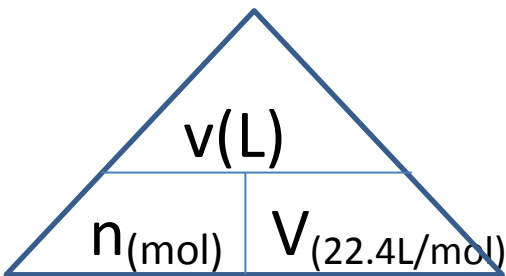


# MOLE WHEEL (molecules, formula units)



$$p = nP$$

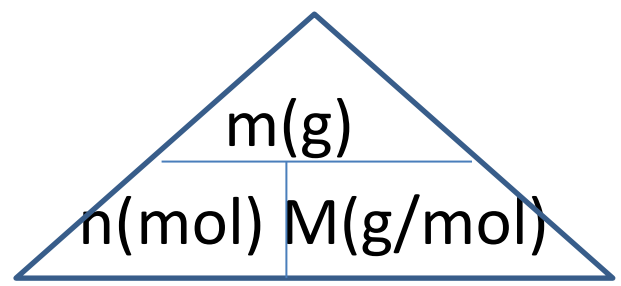


$$v = nV_{STP}$$

$V_{STP} = 22.4 \text{ L/mol}$  (gases only)

$T = 0 \text{ C}$

$P = 101 \text{ kPa}$



$$m = nM$$

$n(\text{mol})$

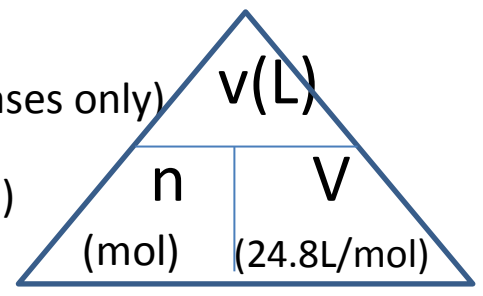
$$n = PV/RT$$

$$v = nV_{SATP}$$

$V_{SATP} = 24.8 \text{ L/mol}$  (gases only)

$T = 25 \text{ C}$  (room temp)

$P = 100 \text{ kPa}$



$$n = CV$$