Chemistry 20 Process Skills, Lab Safety & Lab equipment



• 1) Questioning 2) Proposing Ideas 3) Designing an experiment
4) Observing & Measuring **5)** Processing Evidence 6) Interpretation of Evidence

Step 1: Questioning

Problem: What effect does <u>manipulated</u> <u>variable</u> have on the <u>responding variable</u>.

Variables:

- -<u>Man</u>ipulated: 1st variable changed
- Responding: 2^{nd} variable that reacts to 1^{st}
- Controlled: variables that are kept the same

Step 2: Proposing Ideas

- Background Information: What do you already know?
- Hypothesis: I hypothesis that (<u>answer</u>) because (<u>logical reason based on</u> <u>background</u>). I will test this by ...

Prediction: I predict that (*answer with quantitative data based on background*)

Step 3: Designing Experiments

- ✓ Prelab Calculations: Show any calculations that need to be done before the experiment.
 - Materials & equipment: List the materials and equipment required to do the lab. Include sizes & amounts if known.
- ✓ **Cautions**: State any safety issues.
- Procedure: Make a step by step account of how the experiment is to be done. OR "Look at"

Step 4: Observing and Measuring

Observations: Write down qualitative (qualities) and quantitative (amounts) observations. Include LABELED charts, tables and diagrams if appropriate.

Step 5: Processing Data

 Analysis: Complete any calculations, charts or tables. Create an appropriate graph. Answer questions posed by the data.
 Line Graphing:

Line Graphing:

- Make a title (use pencil & ruler)
- Label & Scale you axis so it covers $\frac{1}{2}$ the page
- Plot your points with circles.
- Draw an appropriate graph (exponential, straight line of best fit, sigmoid or curved)
- Calculate the slope for straight lines(rise/run)
- Add a key if more than two lines present.

Step 6: Interpretation

Conclusion:

- Did the data support the hypothesis, problem, or prediction? Why or why not
- What conclusion(s) are supported by the data? (State the data.)
- What improvements could be made to the experiment?
- What instrumental or human errors may have occurred?

/ Extension:

 What new questions or investigations could be done in the future? (Site references.)

Lab Safety

Everyone is RESPONSIBLE

Safe Practices

- 1) No fooling around.
- 2) Pre-lab preparation is the key to safety
- 3) Wear lab coats and closed shoes.
- 4) Contacts need to be removed. Safety glasses are recommended
- 5) Tie back long hair or loose clothes.
- 6) Remove jewlery.
- 7) Report all accidents

Broken Glass

Is The most common accident in the lab.

If glassware is broken, report it to your teacher immediately.

Follow the teachers instructions when cleaning up the glass.

Cuts & Scrapes

Report all cuts immediately
There is a possibility for infection .
Do not come into contact with another person's blood .

If there is blood go to your seats and wait for instructions.

Chemical Spills

Treat all spills as **dangerous**.

Report spills to your teacher and wait for instructions.

Consult MSDS for special instructions.



Fire

Bunsen burners are the most common source of fires

Turn off the burner when not using them.

If a paper fire occurs, push the paper into the sink and turn on the water.

continued

Fire

Clothing or hair is the most dangerous type of fire.

Cover clothing or hair fires with a fire blanket or use the shower.

The teacher should be the only one to use the fire extinguisher.



Lab Safety

NEVER ENDS

MSDS & WHMIS

Material Safety Data Sheets & Workplace Hazardous Materials Information System

U.S. DEPARTMENT OF LABOR OMB No. 44-B1: Occupational Safety and Health Administration						
MATERIA	1	SAFE	TY DATA	SHEET		
INATENIA	-	UNIL		UNLET		
Required under USE Shipbuilding,	and S	ety and H hipbreakir	lealth Regulations for ig (29 CFR 1915, 19	Ship Repairing, 16, 1917)		
		SECT				
MANUFACTURER'S NAME EMERGENCY TELEPHO						
ADDRESS (Number, Street, City, State, and ZIP C	ode)					
CHEMICAL NAME AND SYNONYMS			TRADE N/	AME AND SYNONYMS		
HEMICAL FAMILY FORMULA						
			1			
SECTION	J 11 -	HAZAF	RDOUS INGREDIE	ENTS		
PAINTS, PRESERVATIVES, & SOLVENTS	*	TLV (Units)	ALLOYS AND METALLIC COATINGS %			(Units)
PIGMENTS			BASE METAL			
CATALYST			ALLOYS			
VEHICLE			METALLIC COATINGS			
SOLVENTS			FILLER METAL PLUS COATING OR CORE FLUX			
ADDITIVES			OTHERS			
OTHERS						TIN
HAZARDOUS MIXTURES OF OTHER LIQUIDS, SOLIDS, OR GASES					~	(Units)
SEC	CTIO	N 111 - P	HYSICAL DATA			
BOILING POINT (°F.)	ING POINT (°F.) SPECIFIC GRAVIT			(H2O=1)		
VAPOR PRESSURE (mm Hg.)	PERCENT, VOLAT BY VOLUME (N)			¢.		
VAPOR DENSITY (AIR=1)	EVAPORATION F			E)		
SOLUBILITY IN WATER						
APPEARANCE AND ODOR						-
SECTION IV	FIR		XPLOSION HAZ	ARD DATA		
FLASH POINT (Method used) FLASH POINT (Method used)						Ue:
EXTINGUISHING MEDIA						
SPECIAL FIRE FIGHTING PROCEOURES						
UNUSUAL FIRE AND EXPLOSION HAZARDS				·.		
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WHMIS SYMBOLS

✓ Class A ✓ Compressed G ✓ ie) acetylene cylinder (welding)



✓ Class B ✓ Flammable & Combustible Material ✓ie) methane







✓ Class D1 Materials causing immediate & serious toxic effects √ie)Acids & Bases







Class D3 Biohazardous Infectious material viruses, biological weapons







Class F Dangerously
 Reactive Material √ie) sodium



Lab Equipment



a. crucible b. evaporating dish c. Watch glass d. Volumetric pipet & graduated pipet e. erlenmeyer flask f. volumetric flask g. funnel

h. wash bottle

i. Stirring rod with rubber policeman j. graduated cylinderk. scoopulal. tongs